CHROM 15 144

KOVÁTS' INDICES OF TRIMETHYLSILYLATED AMINO ACIDS ON FUSED-SILICA CAPILLARY COLUMNS

EWA GAJEWSKI, MIRAL DIZDAROGLU* and MICHAEL G SIMIC

Center for Radiation Research, National Bureau of Standards, Building 245, C 216, Washington DC 20234 (US.4.)

(First received May 10th, 1982; revised manuscript received June 22nd, 1982)

SUMMARY

Trimethylsilyl derivatives of protein amino acids were separated by high-resolution gas chromatography on three fused-silica capillary columns, coated with SE-54, SP-2100 and Carbowax 20M, respectively Kováts' retention indices were also calculated by using *n*-alkanes as standards and tabulated Excellent reproducibility of the index values were obtained The usefulness of the method in actual amino acid analysis of peptides and proteins was demonstrated by analyzing the HCl-hydrolyzate of lysozyme

INTRODUCTION

During the past two decades, a great deal of effort has been made to analyze amino acids by gas chromatography (GC) Amino acids, however, must be derivatized prior to their GC analysis. About 100 different derivatization methods have been suggested and successfully used for this purpose¹⁻³ Zomzely *et al*⁴ introduced the N-trifluoroacetyl (N-TFA) *n*-butyl esters as a possible derivative Later. Kaiser *et al.*⁵ extensively investigated these derivatives for the GC analysis of amino acids Many other N-acyl alkyl ester derivatives have also been utilized for the same purpose⁶⁻¹³ The use of thiohydantoin derivatives in GC has been reported, especially for the sequence analysis of peptides^{14,15}. Trimethylsilylation is another derivatization method, which was introduced by Rühlmann and Giesecke¹⁶. Several authors have used this method for the GC analysis of amino acids utilizing various silylation reagents¹⁷⁻²³

Most of the papers listed above have reported separations using conventional packed columns GC with capillary columns, however, offers great advantages over GC with packed columns²⁴⁻²⁷ This technique has also been applied to GC analysis of amino acids^{7 9 11–13 15 28–31} Several review articles have appeared dealing with GC of amino acids^{1 3 15 32–34}.

In this paper, we describe the GC Kovats' retention indices³⁵ of trimethylsilylated amino acids on three fused-silica capillary columns³⁶ coated with SE-54. SP-2100 and Carbowax 20M, respectively

EXPERIMENTAL*

Apparatus

A Hewlett-Packard Model 5880A microprocessor-controlled gas chromatograph (Hewlett-Packard, Avondale, PA, U.S A) equipped with a flame ionization detector was used. The injection port and detector were maintained at 250°C. Injections were made manually. Helium was used as the carrier gas for all separations at an inlet pressure of 75 kPa. The split ratio was 1°100. The following three fused-silica capillary columns (wall-coated open tubular) (Hewlett-Packard) were used (each 12 m \times 0.2 mm I D.):

(1) SE-54 (5% phenyl, 1% vinyl, methyl silicone gum: silovane deactivated) The measured efficiency was *ca*. 5400 theoretical plates per meter based on the pentadecane peak at 120°C (capacity factor, k' = 6 16; linear velocity $\bar{\mu} = 39.9$ cm/sec). "Trennzahl" (TZ) between tetradecane and pentadecane was 23 3 The ratio of 2,6dimethylphenol (DMP) to 2,6-dimethylaniline (DMA) was 1.00

(2) SP-2100 (methyl silicone fluid, Carbowax 20M deactivated) The measured efficiency was *ca*. 4400 plates per meter based on the pentadecane peak at 130°C (k' = 6.58; $\bar{\mu} = 33.3$ cm/sec). TZ was 200 The ratio of DMP to DMA was 100

(3) Carbowax 20M (polyethylene glycol, Carbowax 20M deactivated) The measured efficiency was ca 5000 theoretical plates per meter, based on the pentadecane peak at 100°C (k' = 2.55; $\bar{\mu} = 33.4$ cm/sec). TZ was 18.7.

Materials

Amino acids were purchased from Vega Biochemicals (Tucson, AZ, USA) Lysozyme was obtained from Sigma (St Louis, MO, USA.). Bis(trimethylsilyl)trifluoroacetamide (BSTFA), acetonitrile and constant-boiling HCl were purchased from Pierce (Rockford, IL, USA) *n*-Alkanes were from Supelco (Bellefonte, PA, U.S.A.).

Trimethy Isily lation

A sample of ca 1 mg of each amino acid was placed in a PTFE-capped hypoval (Pierce) and trimethylsilylated with 0.4 ml of a mixture of BSTFA and acetonitrile (1:1) by heating for 45 min at 140 C.

Hydrolysis of lysozyme with HCl

A sample of ca. 10 mg of the native protein was hydrolyzed with 0 6 ml of constant-boiling HCl in evacuated and sealed tubes at 110°C for 24 h. After hydrolysis samples were dried *in vacuo* and then trimethylsilylated as above

RESULTS AND DISCUSSION**

The Kováts' retention index system³⁵ has generally been accepted as the best way to characterize the retention behavior of chromatographically separated com-

^{*} Certain commercial equipment, instruments or materials are identified in this paper in order adequately to specify the experimental procedure. Such identification does not imply recommendation or endorsement by the National Bureau of Standards, nor does it imply that the material or equipment identified are necessarily the best available for the purpose.

^{**} Abbreviations for amino acids follow IUPAC-IUB recommendations [see Biochem. J., 126 (1972) 773] C10, etc., represents an *n*-alkane with ten carbon atoms pC corresponds to picocoulomb (see Fig. 17)

pounds³⁷⁻³⁹. In most cases, *n*-alkanes are used in GC as standards for this purpose

In the present work, we separated trimethylsilylated protein amino acids by GC on three fused-silica capillary columns and also determined the Kovats' indices of these compounds using *n*-alkanes with an even number of carbon atoms as standards *n*-Alkanes were added to original samples prior to injection. The gas holdup time was determined by using methane, which was injected together with the sample Two silicone stationary phases, SE-54 and SP-2100, and one polar stationary phase. Carbowax 20M (see ref 40 for McReynolds constants of these phases), were used for separation of amino acids and determination of their retention index values. Ideally, symmetrical peak shapes have to be obtained for precise determination of Kovats' indices The Carbowa 20M column, however, did not provide symmetrical peaks for all the amino acids Some of them particularly those with long retention times, vielded tailing peaks. Despite this fact, retention indices of all the amino acids were also determined on this polar column in order to provide sufficient confidence in the identification by matching retention indices. The use of columns of different polarity greatly increases the usefulness of the retention index system for identification purposes Overloading of the columns was avoided, because it causes leading peak shapes, and the position of the peak maximum shifts resulting in incorrect index values

According to definition, Kovats' indices are determined at constant temperature In this paper, we have determined the retention indices of amino acids mainly at constant temperature. It is however, impossible to determine the index values of all the amino acids in a single run at a certain constant temperature, because of the large differences in their retention times. Thus, index values were determined at different constant temperatures corresponding to a reasonable retention time for individual amino acids and an acceptable resolution. Retention indices obtained by using a temperature program were also given

Major error sources in retention index determinations are incorrect temperature measureme is and inaccurate measurement of the retention times. In this paper, the oven temperature was controlled by a microprocessor Retention times were also measured by the microprocessor which reported these values to the nearest 0 001 min^{27} Injections were made manually. Mean retention index values and standard deviations were calculated from at least ten measurements. As the results below show, excellent reproducibility was obtained, particularly on SE-54 and SP-2100 columns

Separations and Kovaty indices

On SE-54 column Four different temperatures, 85° C, 120° C, 150° C and 190° C, were used for measurement of the retention index values of all the amino acids on this column. Chromatograms obtained at these temperatures are given in Figs 1–4, respectively Fig. 5 shows the separation of all the amino acids in a single run by using a temperature program

Figs 1-4 also include *n*-alkanes used for calculation of index values. Mean index values and standard deviations obtained from at least ten measurements are given in Table I. All the amino acids gave symmetrical peaks on this column, providing accurate determination of the index values. Standard deviations varying from 0.03 to 0.12 depending on individual compounds were achieved. Temperature programming yielded slightly higher standard deviations for some amino acids.



Fig. 1. Separation of some trumethyls,lylated amino acids Column, fused-silica SE-54, $12 \text{ m} \times 0.2 \text{ mm}$ I D; temperature 85°C. For other column details see Experimental.



Fig. 2. Separation of some trimethylsilylated amino acids. Column details as in Fig. 1 except temperature 128°C.



Fig 3 Separation of some trimethylsilylated amino acids Column details as in Fig 1 except temperature 150°C



Fig. 4 Separation of some trimethylsilylated amino acids. Column details as in Fig 1 except temperature 190°C.



Fig. 5 Separation of trimethylsilylated amino acids Column details as in Fig 1 except temperature programmed at 4^{-} C/min from 70°C to 170°C, then at 8°C/min from 170°C to 250°C

TABLE I

KOVÅTS' INDICES (1) (MEAN \pm STANDARD DEVIATION) OF AMINO ACIDS ON SE-54 COLUMN

Amuno acid	I _{85°C}	IIMC	I 150°C	I 1997 C	Temp program (Fig 5)
Ala	111027 ± 00^{7}				1119.33 ± 0 07
Val	1227 77 + 0 05				1234.50 ± 0.05
Leu	1287 56 ± 0 06				1299 65 + 0 07
Pro	1304 19 - 0 05				1316 04 ± 0 10
lle	130925 ± 003				$1320\ 60\ \pm\ 0\ 07$
Gly	131687 ± 0.05				1328.30 ± 0.06
Ser	1388.23 ± 0.04				1388 61 ± 0 05
Thr		1404 11 ± 0.05			1416 30 ± 0 07
Arg		1467.67 ± 0 09			1479 89 ± 0 11
Met		1531 10 ± 0.04			1541 58 ± 0 06
Asp		1543 89 <u>+</u> 0 04			1557 09 ± 0 06
Cys		1574.31 <u></u> 0 07			1580 37 ± 0 06
Phe		1635.10 ± 0 09			1640 06 ± 0 13
Arg		1638 81 <u>–</u> 0 10			1646 83 ± 0 06
Glu		1647 08 ± 0 07			1655 51 ± 0 09
Asn		1662.50 ± 0.11			1671.72 ± 0 10
Asn		1697.85 ± 0.06			1707.08 ± 0 06
Gln			1762.35 ± 011		1774 99 ± 0 11
Gln			1794 37 ± 0 11		1800 😑
Arg			-		1848 24 ± 0.17
His			1942.25 ± 0.10	1941 81 <u>+</u> 0 11	1947 08 ± 0 08
Lys			1947 43 ± 0.12	1941.81 ± 0.11	1952.97 ± 0.09
Tyr			1960 08 ± 0 06	1959.22 ± 0 05	1964 36 ± 0 07
Trp				2237.31 ± 0 10	2235 62 <u>+</u> 0 20
Cys-Cys			- .	2327.19 ± 0 09	2332.95 ± 0 17

As and Gln gave two peaks due to trimethylsilylation to different extents²³ (Figs. 2, 3 and 5) Three peaks of Arg (Fig 5) correspond to its decomposition products The third peak represents ornithine The other two decomposition products were not identified.

On SP-2100 column. SP-2100 is a methyl silicone phase whose polarity is slightly less than that of SE-54 (ref. 40)

Retention indices obtained on this column are given in Table II Figs 6-9 show the actual separations at different temperatures Separation of all the amino acids using a temperature program is demonstrated in Fig 10 These chromatograms also show the positions of *n*-alkanes used for index calculations

The elution order of the amino acids is the same as that on the SE-54 column, with one exception. Tyr elutes earlier than Lys (compare Figs 3 and 8, and Figs 5 and 10) On this column, Lys is completely separated from His with a difference of ca. 37 index units at 150°C (Fig 8 and Table II), whereas SE-54 column does not give a complete separation of these amino acids (Fig. 3 and Table I). On the other hand, a better separation of Tyr from Lys was obtained on SE-54 (ca 13 index units *versus ca* 4 index units at 150°C, Tables I and II, respectively, compare also Figs 3 and 8).

Two compounds, Trp and Cys–Cys, gave somewhat broad peaks on SP-2100 column (Figs. 9 and 10) All the other amino acids are represented by symmetrical peaks. A repeatability of 0 04–0.15 index units was achieved for most amino acids on this column (Table II) Higher standard deviations were obtained only for His, Trp (both 0 28) and Cys–Cys (1 00) This is probably due to the fact that these compounds



Fig. 6 Separation of some trimethylsilylated amino acids Column, fused-silica SP-2100 $12 \text{ m} \times 0.2 \text{ mm}$ I D; temperature 90°C For other column details see Experimental



Fig. 7. Separation of some trimethylsilylated amino acids. Column details as in Fig. 6 except temperature 135°C



Fig. 8. Separation of some trimethylsilylated amino acids. Column details as in Fig. 6 except temperature

Fig. 9. Separation of some trimethylsilylated amino acids. Column details as in Fig. 6 except temperature 218 °C.



Fig. 10 Separation of trimethylsilylated amino acids. Column details as in Fig. 6 except temperature programmed at 6° C/min from 80° C to 250° C

TABLE II

KOVÁTS' INDICES (1) (MEAN \pm STANDARD DEVIATION) OF AMINO ACIDS ON SP-2100 COLUMN

Amino acid	I _{90°C}	I _{135°C}	I _{150 C}	I ₂₁₀ .c	Temp.program (Fig 10)
Ala	1106 11 = 0 09				1113.37 ± 0.20
Val	1225 60 ± 0 07				1229 37 - 0 13
Leu	1284 85 ± 0 05				1294.37 ± 0.08
Рго	1296 99 ± 0 05				1308.31 ± 0 14
lle	130652 ± 0.06				131634 ± 0.05
Gly	1315 88 ± 0 04				1326.30 ± 0.20
Ser	1383 95 ± 0 11				1383 44 ± 0 04
Thr		1400	1400		1411 67 ± 0 14
Arg		1448 37 ± 015	1451 46 ± 0 07		1458 22 ± 0 03
Met		1516.78 ± 0.08	1518 48 ± 0 06		1527 51 ± 0 13
Asp		1531 51 ± 0 07	1528 16 ± 0 08		1547 12 ± 0 07
Cys		156323 ± 010	1562 44 ± 0 07		1571.22 = 0.05
Phe		1622.47 ± 0 07	1624 82 ± 0 05		1624.42 ± 0.08
Glu		1634 25 ± 0 15	1629 40 ± 0 07		1642.34 ± 0 10
Asn		$1664 12 \pm 0.06$	1660.13 ± 0.04		1674.29 ± 0.17
Asn		1680 55 ± 0 05	1676 14 ± 0.13		1689.41 ± 0 11
Gin		1773 65 <u>-</u> 0 07	1768.75 ± 0.09		1776.96 ± 0.10
Gln		1784 82 ± 0.15	1780.34 ± 0.14		1784.27 ± 0.06
His			1916 60 ± 0 28	1921 05 ± 0.13	1920 38 ± 0.10
Tyr			1949.42 ± 0 09	1950 38 <u>+</u> 0 08	1954 20 ± 0 08
Lys			1953 89 ± 0 07	1950 38 🛨 0 08	1958 35 <u>+</u> 0 14
Trp			-	221850 ± 0.28	2208 39 ± 0.27
Cys-Cys				2314.22 ± 1 00	2320 37 ± 0.41



Fig. 11. Separation of some trimethylsilylated amino acids. Column, fused-silica Carbowax 20M $12 \text{ m} \times 0.2 \text{ mm}$ I D., temperature 75°C. For other column details see Experimental.



Fig. 12. Separation of some trimethylsilylated amino acids. Column details as in Fig. 11 except temperature 135°C.

Fig. 13. Separation of some trimethylsilylated amino acids Column details as in Fig. 11 except temperature 150°C.



Fig 14 Separation of some trimethylsilylated amino acids Column details as in Fig 11 except temperature 200°C

TABLE III

Amino acid	I _{75 C}	I _{135 C}	I _{160 C}	I _{260 C}
Ala	$1206\ 10\ \pm\ 0\ 10$			
Val	1289 72 ± 0 07			
Leu	1358 58 ± 0 12			
Gly	1364 90 ± 0 03			
Ile	1369 55 ± 0.05			
Pro	1396 61 🛨 0 07			
Thr	1469 26 - 0 09			
Ser	147609 ± 006			
Asn		1653 32 ± 0 15		
Asp		168385 ± 014		
Cys		$1703 19 \pm 0.06$		
Arg		1726 72 ± 0 20		
Met		$1741\ 00\ \pm\ 0\ 16$		
Gln		$1786\ 61\ \pm\ 0\ 28$		
Glu		1800		
Phe		1863 65 ± 0 11	1863 83 ± 0 07	
Lys		1971 52 ± 0 16	$1944\ 21\ \pm\ 0.10$	
Asn		—	202871 ± 013	
Туг			2151 59 ± 0 15	
Gin			$216540 \div 020$	
His				2293 85 + 0 91
Cvs-Cvs				253721 ± 078
Тгр				254852 ± 026

KOVÁTS' INDICES (I) (MEAN \pm STANDARD DEVIATION) OF AMINO ACIDS ON CARBOWAX 20M COLUMN

~



Fig. 15 Separation of some trimethylsilylated amino acids. Column details as in Fig. 11 except temperature programmed at 5°C/min from 70°C to 200°C.



Fig. 16. Separation of some amino acids at the highest sensitivity setting of the detector. Column details as in Fig. 5. Amount injected for each amino acid was *ca*. 0.6 pmol.



yielded broad and tailing peaks on this column. Standard deviations were slightly higher for some amino acids when temperature programming was used.

On Carbowax 20M. As expected, this highly polar column has a totally different selectivity from the other two columns toward trimethylsilyl (TMS) derivatives of amino acids as Figs 11–14 and Table III clearly demonstrate The majority of the amino acids gave "tailing" peaks. As mentioned above, symmetrical peaks are required for accurate determination of index values. Nevertheless, index values of the amino acids separated on this column were also calculated and tabulated (Table III)



Fig. 18 Separation of the HCl-hydrolyzate of lysozyme Column details as in Fig. 1 except temperature programmed at 4°C/min from 70°C to 250°C

An additional column with a different polarity and selectivity can, of course, be very useful for identification purposes Standard deviations were higher on this column than those on SE-54 and SP-2100 columns. Fig. 15 shows the separation of some amino acids using a temperature program

Sensitivity

The method described here provides a high sensitivity. Fig. 16 shows the separation of six amino acids at the highest sensitivity setting of the detector. Each peak represents ca. 0.6 pmol of an amino acid. A signal-to-noise ratio of 2 (minimum detection limit) corresponds to ca. 0.1 pmol. Detector response for all amino acids was found to be linear from 0 l to 20 pmol injected as shown in Fig. 17 for Ser

CONCLUSIONS

As the results above show, the SE-54 column is best suitable for separation of TMS derivatives of the amino acids among the columns tested here All the amino acids including the most troublesome ones, such as Arg, His, Lys, Trp and Cys–Cys, gave symmetrical peaks. Also, the best reproducibility of the index values was achieved on this column. The superior performance of the SE-54 column in comparison to the SP-2100 column may be due to the type of column deactivation (see Experimental) or simply to the slightly higher polarity of the SE-54 stationary phase⁴⁰.

In addition, we recommend the use of short (12 m), narrow-bore (0.2 mm I D) fused-silica capillary columns for GC analysis of trimethylsilylated amino acids. According to our experience, TMS derivatives of amino acids, particularly those with long retention times, tend to decompose on longer (25 m) columns. This is particularly potentiated for troublesome amino acids such as Arg, His, Lys, Trp and Cys-Cys. These compounds decompose either partly or completely on 25-m SE-54 and SP-2100 columns. No peaks were obtained on a 25-m Carbowax 20M column

The usefulness of the methodology described here is clearly demonstrated in Fig 18 where the separation of the HCl-hydrolyzate of lysozyme is shown. Because of its excellent resolving power and very high sensitivity, this method could be well applied to amino acid analysis of peptides and proteins

REFERENCES

- 1 K Blau, in H. A. Szymanski (Editor), Biomedical Applications of Gas Chromatography, Vol 2, Plenum, New York, 1968 p 1.
- 2 P Hušel and K. Macel, J. Chromatogr , 113 (1975) 139
- 3 K Blau, in J M Rattenbury (Editor), Amino Acid Analysis, Wiley, New York, 1981, p 48
- 4 C. Zomzely, G. Marco and E. Emery Anal Chem., 34 (1962) 414
- 5 F. E. Kaiser, C. W Gehrle, R. W. Zumwalt and K. C. Kuo. J Chromatogr , 94 (1974) 113
- 6 C W Moss, M. A. Lambert and F. J Diaz, J Chromatogr 60 (1971) 134
- 7 J C. Cavadore, G Nota, G Prota and A. Previero, Anal Biochem, 60 (1974) 608
- S. M. Makita, S. Yamamoto and M. Kono, J. Chromatogr., 120 (1976) 129
- 9 C F. Poole and M Verzele, J Chromatogr, 150 (1978) 439
- 10 S. L. MacKenzie and D Tenaschuk, J. Chromatogr , 171 (1979) 195
- 11 J Desgres, D Boisson and P. Padieu, J. Chromatogr., 162 (1979) 133
- 12 R. F Adams, F. L. Vandemark and G J. Schmidt, J Chromatogr Sci., 15 (1977) 63
- 13 H. Frank, G. J. Nicholson and E. Bayer, J. Chromatogr., 167 (1978) 187.
- 14 J J Pisano, T. J Bronzert and H. B Brewer, Jr., Anal Biochem, 45 (1972) 43.

- 15 Z. Deyl, J Chromatogr 127 (1976) 91
- 16 K Rühlmann and W Giesecke, Angew Chem 73 (1961) 113
- 17 E D Smith and H Sheppard Jr Nature (London) 208 (1965) 878
- 18 D L Stalling, C W Gehrke and R W Zumwalt Biochem Biophys Res Commun. 31 (1968) 616
- 19 C W Gehrke, H Nakamoto and R. W Zumwalt, J Chromatogr 45 (1969) 24
- 20 K Bergstrom, J Gurtler and R Blomstrand Anal Biochem, 34 (1970) 74
- 21 C W Gehrke and K Leimer J Chromatogr 53 (1970) 201
- 22 R W Zumwalt K Kuo and C W Gehrke, J Chromatogr 57 (1971) 193
- 23 C W Gehrle and K Leimer J Chromatogr 57 (1971) 219
- 24 G Schomburg H Husmann and F Weeke J Chromatogr 99 (1974) 63
- 25 G Schomburg and H Husmann Chromatographia 8 (1975) 517
- 26 M Novotny, Anal Chem , 50 (1978) 16A
- 27 R R Freeman High Resolution Gas Chromatography Hewlett-Packard Palo Alto CA 1979
- 28 J Johnson J Eyem and S Sjoquist Anal Biochem 51 (1973) 204
- 29 R J Pearce J Chromatogr 136 (1977) 113
- 30 H Frank G J Nicholson and E Bayer J Chromatogr Sci 15 (1977) 174
- 31 I Abe T Kohno and S Musha Chromatographia 11 (1978) 393
- 32 J J Pisano Methods Enzymol 25 (1972) 27
- 33 W Jennings Gas Chromatography with Glass Capillary Columns Academic Press New York 1980 p 231
- 34 H Jaeger, H Frank H U Klor and H Ditschuneit in W Jennings (Editor) Glass Capillari Gas Chromatographi -- The Applictions Marcel Dekker New York 1980
- 35 E Kovats Advan Chromatogr 1 (1965) 229
- 36 R D Dandeneau and E H Zerenner J High Resolut Chromatogr Chromatogr Commun 2 (1979) 351
- 37 G Schomburg and G Dielmann, J Chromatogr Sci 11 (1973) 151
- 38 L S Ettre Chromatographia 6 (1973) 489
- 39 L S Ettre Chromatographia 7 (1974) 39
- 40 W O McReynolds, J Chromatogr Sci 8 (1970) 685